Glycerol-¹³C₃

MedChemExpress

Cat. No.:	HY-B1659S5	
CAS No.:	63346-81-6	
Molecular Formula:	¹³ C ₃ H ₈ O ₃	ОН НО\ ¹³ СН _ОН
Molecular Weight:	95.07	
Target:	Endogenous Metabolite	HO ¹³ CH OH
Pathway:	Metabolic Enzyme/Protease	H_2 H_2
Storage:	4°C, sealed storage, away from moisture and light * In solvent : -80°C, 6 months; -20°C, 1 month (sealed storage, away from moisture and light)	

BIOLOGICAL ACTIVITY		
Biologicke Activity		
Description	Glycerol- ¹³ C ₃ is the ¹³ C labeled Glycerol[1]. Glycerol is used in sample preparation and gel formation for polyacrylamide gel electrophoresis[2][3][4].	
In Vitro	Stable heavy isotopes of hydrogen, carbon, and other elements have been incorporated into drug molecules, largely as tracers for quantitation during the drug development process. Deuteration has gained attention because of its potential to affect the pharmacokinetic and metabolic profiles of drugs ^[1] . MCE has not independently confirmed the accuracy of these methods. They are for reference only.	

REFERENCES

[1]. Russak EM, et al. Impact of Deuterium Substitution on the Pharmacokinetics of Pharmaceuticals. Ann Pharmacother. 2019 Feb;53(2):211-216.

[2]. Pennings S, et al. Effect of glycerol on the separation of nucleosomes and bent DNA in low ionic strengthpolyacrylamide gel electrophoresis. Nucleic Acids Res. 1992 Dec 25;20(24):6667-72.

[3]. Yazdani SS, et al. Anaerobic fermentation of glycerol: a path to economic viability for the biofuelsindustry. Curr Opin Biotechnol. 2007 Jun18(3):213-9.

[4]. Huang ZH, et al. Expression and function of P-glycoprotein in rats with glycerol-induced acute renal failure. Eur J Pharmacol. 2000 Oct 20406(3):453-60.

Caution: Product has not been fully validated for medical applications. For research use only.

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Product Data Sheet